

PREPARATION OF MACRO- POROUS ALUMINA VIA ORGANIC ADDITIVE AND CHARACTERIZATIONS PHYSICAL PROPERTIES BY USING A GENETIC ALGORITHM METHOD

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Abstract-The exhibited work is focused on the preparation of alumina based porous ceramics and its characterizations. Alumina ceramic with Macro - porous size and control of pore morphology was prepared by semi-dry pressing method via using organic yeast cell for pore forming. Where, alumina powder synthesis by sol-gel method using $Al(NO_3)_3 \cdot 9H_2O$ and ammonia, then it uses as raw material to produce a porous alumina bodies. During the heat treatment, the organic additions(yeast) was removed and produce a pore ,it was observed that with increasing amount of the yeast cell, the porosity tends to increase. The effect of both different mas ratio of alumina synthesis to yeast, pressing, calcination temperature and soaking time on the microstructure of the final bodies were an investigation. The obtain macro porous alumina had a pore size in rand (0.09-2 μm), porosity (53-75%).

Keywords: Alumina, Yeast, Sol-gel process, Semi-dry pressing, Genetic algorithm method

Yeast cells were employed in the synthesis of alumina with combined macro-/meso porosity. This is because the solid cell wall structure of yeast whose hardness is perfect it was interesting to withstand the pretreatment procedure [7].

GA is regarded as one of the most successful meta-heuristic techniques for solving combinatorial optimization problems. Thus, a system using GA produces a variety of(individuals) and selects the fittest according to criteria, change the individual, and repeat the process on the next generation, GA works on an encoding of the parameter set rather than the parameter set itself[8-9].

In view of that, the objective of this research is to macro porous alumina by using alumina and yeast cell as starting materials characterizations physical properties by using a Genetic Algorithm Method

I. INTRODUCTION

Porous alumina has a wide use in many industrial field, such as catalyst supports, heat exchangers, thermal insulation, filtration liquid or metal molten, hot-gas purifier and in the biomedical because of their properties such as low density, low thermal conductivity, high chemical stability, resistance to high temperature, large surface area [1-2] . Already several processing techniques were used to prepare of the porous alumina ceramics, such as partial sintering method [3], gel casting, freezing casting [4], organic foam method [5] and pore-forming agent method [6]. Among these methods, pore-forming agent technique is the most common and highly effective technique. There is many pore-forming agents including yeast cell, starch, graphite, and naphthalene and rice husk.

II. EXPERIMENTAL

A. ALUMINA PREPARATION

50 ml of Aluminum nitrate ($Al(NO_3)_3 \cdot 9H_2O$) was dissolved in deionized water under magnetic stirring. Then, the ammonium hydroxide solution was added as a droplet to the mixture. The initial aluminum/ammonia solution pH was 2 gradually increased and then rose up to 8.8 under continues stirring. The mixture was stirred continually at 40 C for 5 h, and then aged for 24 h at room temperature resulting convert the sol to gel. The gel was then filtered and washed with deionized water. The freshly prepared alumina gels was dry at 150°C for 6 h, and resulted amorphous Boehmite-alumina powder.

B. MACRO POROUS ALUMINA PREPARATION.

The macro-porous alumina was prepared by mixing boehmite alumina powder (2 μm) with yeast cell as pore-forming agent at different weight percent ratios (5%, 10%, 15%, and 20%). The study was added 2% polyvinyl alcohol (PVA) to bind the mixture and formed using semi-dry pressing method by compacting the powder in steel die of diameter (20 mm) by compression hydraulic device at different compaction pressure. The samples are then dried by heating at 100 °C for 4 and then sintering in different temperature (1200, 1300, 1400, 1500).

C. PHYSICAL PROPERTIES

1) porosity

The specimens were tested by Archimedes method according to (ASTM C373-88). The porosity is relationship between the open pores to exterior volume and can be determined as following:

$$p = \left(\frac{M - D}{V} \right) * 100 \quad \dots\dots (1)$$

Where:

$$v = m - s$$

- M= Saturated weight
- D= dry weight
- S= Suspense weight

2) Bulk Density

The bulk density of specimens is calculated after sintering according to ASTM (373-88) by the following equation

$$\rho = D / v \quad \dots\dots (2)$$

III. RESULT AND DISCUSSION

A. XRD RESULTS

The phases of prepared powders were identifying using XRD (Shimadzu 6000, Japan, Department of Ceramics Engineering and Building Materials, University of Babylon) in room temperature using Cukα radiation. Figure 1.a,b exhibits the result of X-ray diffraction testing (a) shows the result of x-ray diffraction analysis alumina powder before heat treatment (b) after heat treatment at above 1200,

As we can see α- Al2O3 was the only phase present for the powder heat treatment above 1200°C.

B. ENERGY DISPERSIVE SPECTROSCOPY (EDS)

EDS was utilized to identify the chemical composition of an element under SEM (Bruker XFlash 630 EDS, pharmacy . university of Babylon), The existence of Al and O with weight percent which confirms by using EDS .EDS shows peaks of aluminum and oxygen and indicates fewer impurities in prepared Al2O3.

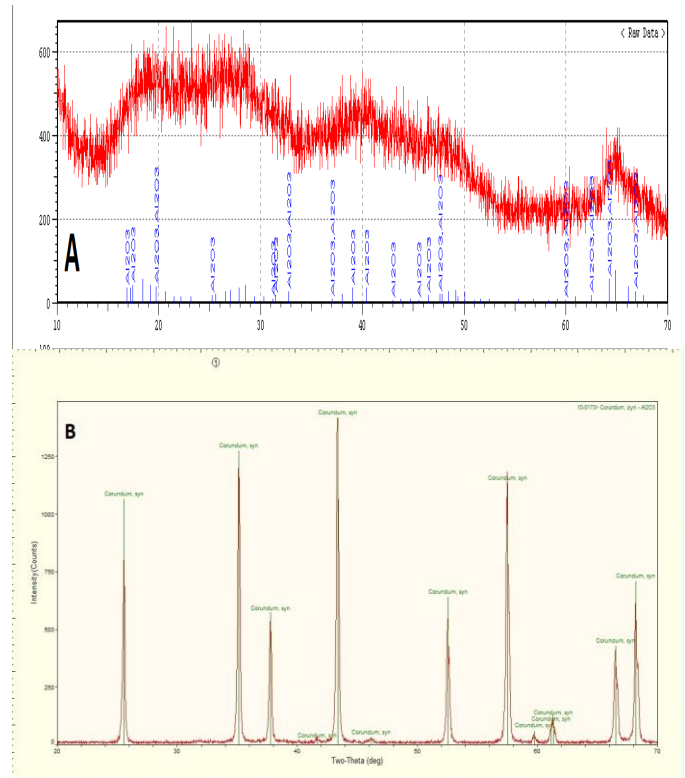


Fig. 1XRD pattern of (a) shows the result of x-ray diffraction analysis alumina powder before heat treatment, (b) after heat treatment at above 1200

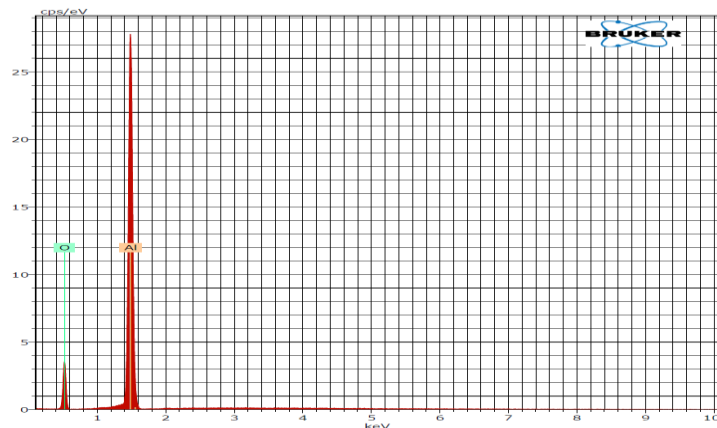


Figure.2 show Energy dispersive spectroscopy (EDS) of Al2O3
El AN Series un. C norm. C Atom. C Error (1 Sigma)
[wt.%] [wt.%] [at.%] [wt.%]

Al	13 K-series	72.55	74.62	63.55 3.39
O	8 K-series	27.45	25.38	36.45 3.23
Total:		100.00	100.00	100.00

C. SEM RESULTS

The microstructure of the alumina based porous ceramics was observed using (SEM) instrument (FEI, America, pharmacy. university of Babylon). Figure.3 (a) SEM shown micrographs of the porous alumina result from alumina with5 % of yeast.

After sintering at 1400 °C for 2h ,(b) with 10 % yeast after sintering at 1400 °C for 2h.,(c) with 5% of yeast after sintering at 1200 °C . it show that when increased the amount of yeast it will be increase the porosity, it may be seen there are smaller pores with the size in range (90-400) nm that are generated by the variability of yeast cell.

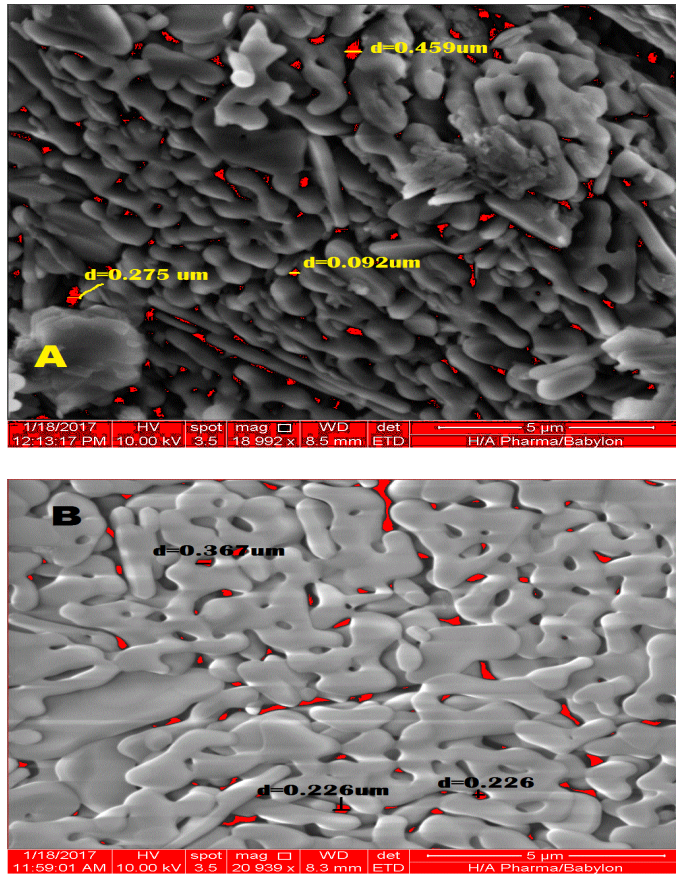


Fig.3 (a, b) Shows SEM micrographs of the porous alumina

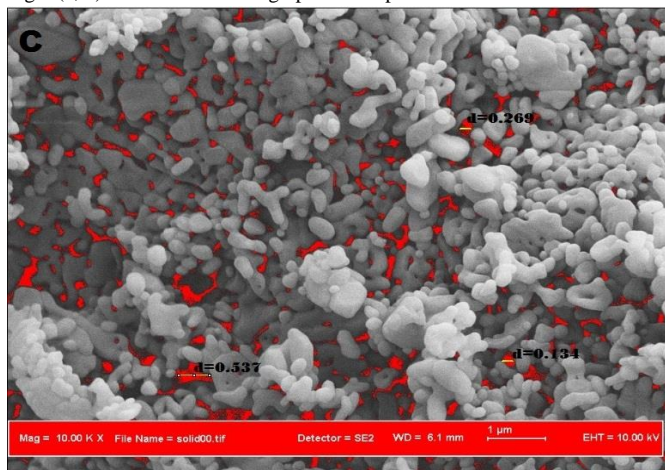


Fig.3 (C) Shows SEM micrographs of the porous alumina

* Red color in a scanning electron microscope images represent the presence of pores

D. PARTICLE SIZE ANALYZER

The particle size distribution was determined by using Bettersize2000 laser particle size analyzer (Better size instrument Ltd., China) in the ceramic laboratories / College of Material Engineering/ Babylon University). Figure .4a,b show the result of analysis of particles size distribution for the alumina powder ,(a) before the calcination, (b) show the result of alumina powder after calcination at 1200 °C.

It can be observed that when increased the temperature of calcination it will increased the particles size of the powder because at higher temperature the growth of particles will be quicker.

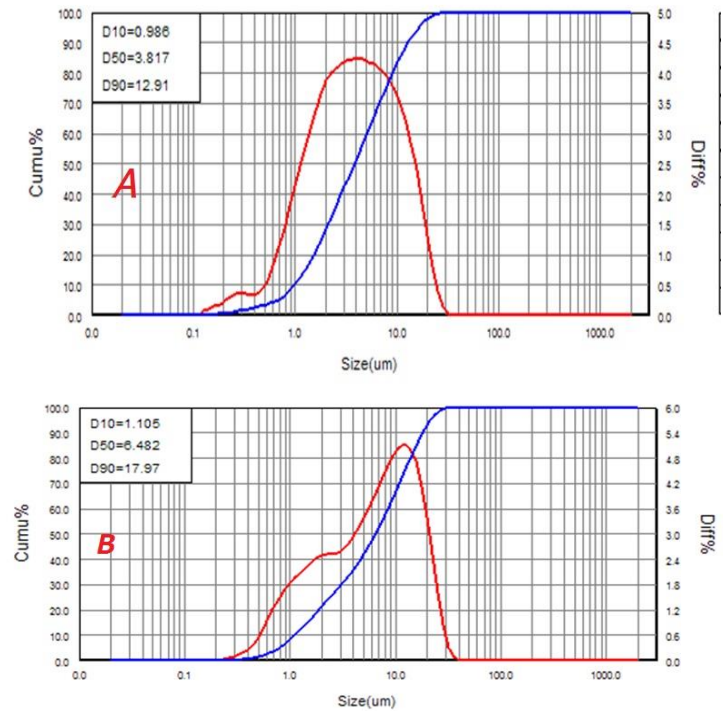


Fig .4. a, b show the result of analysis of particles size distribution for the alumina powder

E. REGRESSION EQUATION ANALYSIS

The prediction model for porosity and density is established by taking porosity as dependent variable and input variables (temperature, soaking time, concentration of the yeast, pressing) as independent variables. Therefore, the design of experiment has used based on 60 experiments for four factors and four levels. Using Minitab software, a regression relationship is established between them,

1) GIVEN IN EQUATION FOR POROSITY

$$\begin{aligned}
 'F(x) = & 7.8 + 0.0329 x(1) + 0.53 x(2) - 26 X(3) + 8.5 X(4) - \\
 & 0.00031 x(1)*x(2) + 0.028 x(1)*X(3) - 0.0031 x(1)*X(4) - \\
 & 1.53 x(2)*X(3) - 0.509 x(2)*X(4) - 34.6 X(3)*X(4) + \\
 & 0.00107 x(1)*x(2)*X(3) + 0.000344 x(1)*x(2)*X(4) + \\
 & 0.0248 x(1)*X(3)*X(4) + 1.82 x(2)*X(3)*X(4) - \\
 & 0.00138 x(1)*x(2)*X(3)*X(4).
 \end{aligned}$$

2) GIVEN IN EQUATION FOR DENSITY

$$F(x) = -130 + 0.145 x(1) + 4.24 x(2) + 13.9 x(3) + 53.5 x(4) - 0.00291 x(1)*x(2) - 0.0100 x(1)*x(3) - 0.0411 x(1)*x(4) - 0.417 x(2)*x(3) - 0.87 x(2)*x(4) - 1.03 x(3)*x(4) + 0.000287 x(1)*x(2)*x(3) + 0.00057 x(1)*x(2)*x(4) + 0.00071 x(1)*x(3)*x(4) + 0.066 x(2)*x(3)*x(4) - 0.000042 x(1)*x(2)*x(3)*x(4)$$

Where:

F(x): porosity

F(x): density

X (1): Temperature

X (2): pressing

X (3): concentration of yeast

X (4): socking time

The Objective Function and Parameter of Genetic Algorithm. The genetic algorithm option in the optimization tool of Matlap software was used to represent solution of the optimization problem. Objective value result in Matlap software, which was (76.19%) for maximum porosity and (49.31%) for minimum density

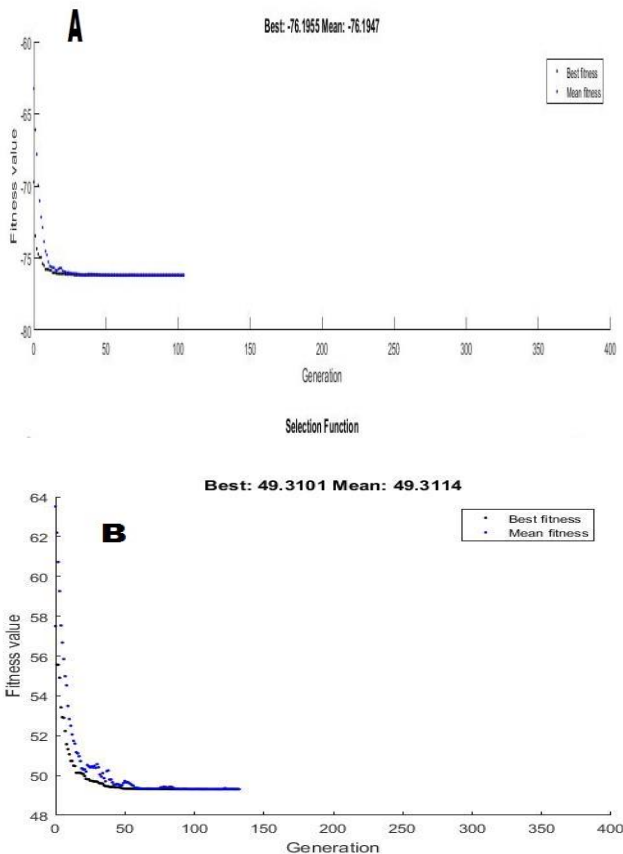


Figure 5. A, b shows the fitness value versus generation

IV. CONCLUSIONS

Macro-porous alumina was successfully prepared using semi-dry pressing method, utilizing natural yeast cell as pore

forming agent The porosity obtained was in the range 53-75% and it increased as the percentage ratio of yeast cell increased. The genetic algorithm optimization was used to optimize the problem parameters, which are involved (temperature, socking time, concentration of yeast and pressing). The maximum porosity using genetic algorithm was 76.195, in approximate agreement with experimental value 75 .and for the density obtained in range (50-77) the genetic algorithm optimization for minimum density id 49.31, in approximate agreement with experimental value 50.1.

V. FUTURE WORK

A study of the mechanical, thermal properties and optimize them by using genetic algorithm and neural artificial network to complete the present work

Macro-porous alumina will prepare by using Natural yeast cell via slip-casting methods. To achieve this point, it will study of mechanical, physical and thermal properties. Then, optimize them by using genetic algorithm and neural artificial network.

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