Ministry of Higher Education and Scientific Research Scientific Supervision and Scientific Evaluation Apparatus Directorate of Quality Assurance and Academic Accreditation Accreditation Department



# Academic Program and Course Description Guide

# **Introduction:**

The educational program is a well–planned set of courses that include procedures and experiences arranged in the form of an academic syllabus. Its main goal is to improve and build graduates' skills so they are ready for the job market. The program is reviewed and evaluated every year through internal or external audit procedures and programs like the External Examiner Program.

The academic program description is a short summary of the main features of the program and its courses. It shows what skills students are working to develop based on the program's goals. This description is very important because it is the main part of getting the program accredited, and it is written by the teaching staff together under the supervision of scientific committees in the scientific departments.

This guide, in its second version, includes a description of the academic program after updating the subjects and paragraphs of the previous guide in light of the updates and developments of the educational system in Iraq, which included the description of the academic program in its traditional form (annual, quarterly), as well as the adoption of the academic program description circulated according to the letter of the Department of Studies T 3/2906 on 3/5/2023 regarding the programs that adopt the Bologna Process as the basis for their work.

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In this regard, we can only emphasize the importance of writing an academic programs and course description to ensure the proper functioning of the educational process.

### **Concepts and Terminology:**

<u>Academic Program Description</u>: The academic program description provides a brief summary of its vision, mission and objectives, including an accurate description of the targeted learning outcomes according to specific learning strategies.

**Course Description**: Provides a brief summary of the most important characteristics of the course and the learning outcomes expected of the students to achieve, proving whether they have made the most of the available learning opportunities. It is derived from the program description.

**<u>Program Vision</u>**: An ambitious picture for the future of the academic program to be sophisticated, inspiring, stimulating, realistic and applicable.

**Program Mission:** Briefly outlines the objectives and activities necessary to achieve them and defines the program's development paths and directions.

**<u>Program Objectives</u>**: They are statements that describe what the academic program intends to achieve within a specific period of time and are measurable and observable.

<u>Curriculum Structure</u>: All courses / subjects included in the academic program according to the approved learning system (quarterly, annual, Bologna Process) whether it is a requirement (ministry, university, college and scientific department) with the number of credit hours.

**Learning Outcomes:** A compatible set of knowledge, skills and values acquired by students after the successful completion of the academic program and must determine the learning outcomes of each course in a way that achieves the objectives of the program.

<u>Teaching and learning strategies</u>: They are the strategies used by the faculty members to develop students' teaching and learning, and they are plans that are followed to reach the learning goals. They describe all classroom and extra-curricular activities to achieve the learning outcomes of the program.

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### **Academic Program Description Form**

University Name: University of Babylon Faculty/Institute: College of Pharmacy Scientific Department: Pharmaceuticals Academic or Professional Program Name: Sciences in pharmacy Final Certificate Name: Bachelor's Academic System: two semesters in each academic year Description Preparation Date: 2023- 2024 File Completion Date:

Signature: Head of Department Name: Signature: Scientific Associate Name:

Date:

Date:

The file is checked by:

Department of Quality Assurance and University Performance Director of the Quality Assurance and University Performance Department: Date: Signature:

Approval of the Dean

1. Course Name:

Physical pharmacy

2. Course Code:

2 / 22 🔀

3. Semester / Year:

Second semester/ second year

4. Description Preparation Date:

17/2/2024

5. Available Attendance Forms:

Class attendance

6. Number of Credit Hours (Total) / Number of Units (Total)

45credit hours /4 units

7. Course administrator's name (mention all, if more than one name)

Assit. Prof. Asmaa Hashim Hammadi <u>phar.asmaa.hashim@uobabylon.edu.iq</u> Prof. Fatma al-zhra jabar

Assit. Prof. Mohammed Obies

8. Course Objectives

Course Obje	ectives	1-Identifying all types and forms of medicines.
		2 - Methods of preparing active ingredients in full dosage formsFor humans and animals
		3 Study the stability of prepared doses in various forms
		4 - Study the drug effect, effectiveness, and mechanism of action within the body.
9. Tea	ching and Learning Strategies	
Strategy	1- Theoretical lectures	
2 - Educational laboratories		
	ch	
	4- Desk research	

- 0	ourse St				
Week	Hours	Required Learning	Unit or subject	Learning	Evaluation
		Outcomes	name	method	method
1-4	12	Solubility and Distribution Phenomena		Lectures, discussions, seminars	Writing exam, oral evaluation
1-3	9		Kinetics, rate and orders of reactions, influence of temperature and other	Lectures, discussions, seminars	Writing exam, oral evaluation
1-3	9		factors on reactions rate, decomposition of medicinal agents and accelerated stability analysis.	Lectures, discussions, seminars	Writing exam, oral evaluation
1-4	12	Colloidal Dispersions		Lectures, discussions, seminars	Writing exam, oral evaluation
1-3	13	Rheology		Lectures, discussions, seminars	Writing exam, oral evaluation

11. Course Evaluation					
The pursuit grade is divided into 20% for theory, 20% for practical, and 60% for the end- of-semester exam					
12. Learning and Teaching Resources					
Required textbooks (curricular books, if any)	1. Martin's Physical Pharmacy and				
Main references (sources)	Pharmaceutical Sciences, 6th edition 2. Skoog, Skoog, Fundamentals of				
Recommended books and references	<b>U</b> ·	cal Chemistry, 9t			
(scientific journals, reports)					
Electronic References, Websites					

1. Course Name: Bio Pharmacy				
2. Course Code: PhBi4B00044				
3. Semester / Year: semester(15 weeks)				
4. Description Preparation Date: 17	//2/ 2024			
5. Available Attendance Forms: week	ly lectures			
6. Number of Credit Hours (Total) / N	lumber of Units (Total)			
3 hours, 2 units				
· · · · · · · · · · · · · · · · · · ·				
7. Course administrator's name (m				
Assist. Prof. Dr. Dhafir Kahtan Sae Assist. Prof. Dr. Aymen Amer Bash				
Assist. 1101. DI. Ayliten Amer Dasi	1			
8. Course Objectives				
Course Objectives	<ol> <li>Knowledge of biopharmaceutical consideration of drugs .</li> <li>Understanding of pharmacokinetie behavior of drugs.</li> <li>biological effectiveness of the drugs inside the human body and quantification of drug concentration in normal subjects and in certain.</li> </ol>			
9. Teaching and Learning Strategies				
scientific lectures at the same easy to understand and unders	lemic competence of students through e level as students in the medical group, stand, and equipping the halls to facilitate nnologies and creating an educational			

environment for the new generation. 2- Improving the educational environment by making it enjoyable and free of anything that causes anxiety for students. This is done by enriching cultural and sports activities and encouraging club activities.

#### 10. Course Structure

	. Course S				
Wee	Hours	Required Learning	Unit or subject	Learning	Evaluation
		Outcomes	name	method	method
1	1	Introduction to biopharmaceutics		Theoretical lectures using projectors and smart boards+ Electronic platform (blended learning) + video lectures	Conducting oral exams after the lecture +Quick surprise exams Quiz + Practical reports +Homework + A monthly theoretical and practical exam + Seminars
2-4			Biopharmaceutic aspects of products; drug absorption; mechanisms of absorption; physicochemical factors; dissolution rate; effects of excipients; type of dosage forms		+ Seminars
5			One compartment open model		
6			Multicompartment models		
7			Pharmacokinetics of drug absorption		

Assist. Prof. Dr. Aymen Amer Course Name: Bio Pharmacy University of Babylon / Collage of Pharmacy									
Assist. Prof. Dr. Dhafir Kahtan Saeed									
(scientific journals, reports) Electronic References, Websites			Biopharm	Shargel L, Yu AB, (Eds.), Applied Biopharmaceutics and Pharmacokinetics 6 <sup>th</sup>			ics 6 <sup>th</sup>		
	commende		and references		. /		\ <b>^</b>		
Ma	ain reference	ces (sources)			Applied Clinical Pharmacokinetics , Latest				t
Re	quired text	books (curric	ular books, if any)	Biopharm	nacy				
1	2. Learr	ning and Te	aching Resources	5					
	•		100 according to the table, or written exams,	•			such as	daily	
	-	se Evaluatio							
			I						
			renal disea						
15			Dosage ad	iustment in					-
14			pharmacok	inetics					
14			Non-linear						
13			Multiple do regimens	sage					
12			Intravenou	s infusion					
11				ding of drugs					
10			Hepatic eliı drugs	nination of					
9			Clearance from the bio systems	ological					
8			Bioavailabi bioequivale						

	Course Description Form					
1. Course Nan	ne:					
Pharmaceutic	Pharmaceutical Technology					
2. Course Code:						
PhPtii3B000	035 (3+2)					
3. Semester /	Year:					
2023-2024						
4. Description	Preparation Date:					
14/2/2024						
	ttendance Forms:					
attendance						
	Credit Hours (Total) / Number of Units (Total)					
45 hours/ 4 units						
7. Course adu	ministrator's name (mention all, if more than one name)					
Name:						
Ghada Hamid						
Naji						
Email:						
phar.ghadah.						
hamid@uoba						
bylon.edu.iq						
8. Course Obje	ctives					
Course Objectives	• Define the various types of solid, liquid and semisolid dosage forms.					
	• List the advantages and disadvantages of these dosage forms in extemporaneous compounded prescriptions and in patient therapy					
	• Define solubility and describe how different factors increase or decrease solute solubility in a given solvent					
	<ul> <li>Evaluate and select a proper solvent and delivery system for a given solute, purpose, and/or patient population</li> </ul>					
	<ul> <li>Identify the challenges of using these various types of dosage forms.</li> </ul>					
	<ul> <li>select the best dosage form according to drug itself and according to disease state of the patient being treated.</li> </ul>					
	<ul> <li>Explain how a drug's physical characteristics influences the pharmaceutical dosage forms that will be used to administer it.</li> </ul>					
	<ul> <li>Compare and contrast liquid dosage forms to solid oral dosage forms and to semisolid dosage forms</li> </ul>					

9.	9. Teaching and Learning Strategies							
Strateg		Basic Thinking Methods High thinking Development of self-capacity Brainstorming						
10. C	course	Structure						
Week	Hou	Required Lea	arning	Unit or subject	Learning	Evaluation		
				name	method			
		Outcomes				method		
1	3	Emulsions	disadv	the advantages and antages of using liquid torms in	Lectures Boards Slide show	Exams seminars		
2	3	Stability of emulsions	extemporaneous compounded prescriptions and in patient therapy 3. Compare and contrast liquid dosage forms to solid oral dosage forms 4. Define solubility and describe how different factors increase or decrease solute		extemporaneous compounded prescriptions and in patient therapyLectures Boards3. Compare and contrast liquid dosage forms to solid oral dosage formsSlide show4. Define solubility and describe how different factorsSlide show		Boards	Exams seminars
3	3	Suppositories	1.Compare and contrast various suppository, insert, and stick dosage forms in terms of		Boards Slide show	Exams Seminars		
4	3	Preparation of suppositories			Lectures Boards Slide show	Exams Seminars		

			preparation		
5	3	Semisolid dosage forms	1.Differentiate between the various types of semisolid bases on the basis of physical and	Lectures Boards Slide show	Exams Seminars
6	3	Preparation of ointments	chemical properties 2. List the criteria for the selection of a semisolid base to treat a topical affliction 3. Describe the methods to incorporate (an) active ingredient(s) into a semisolid base 4.Explain the difference between an ointment, a cream, and a gel. 5. List advantages and disadvantages of administering drugs rectally and vaginally	Lectures Boards Slide show	Exams Seminars
7	3	Powders	<ol> <li>Differentiate a powder from a granule</li> <li>Explain how a drug's powder particle size influences the pharmaceutical dosage forms that will be used to administer it</li> <li>Define micromeritics, the angle of repose, levigation, spatulation, and trituration</li> </ol>	Lectures Boards Slide show	Exams Seminars
8	3	Granules	1.Define granules 2. Differentiate between the fusion method and wet method for the Preparation of effervescent granulated salts	Lectures Boards Slide show	Exams Seminars
9	3	Hard gelatin capsules	1.Differentiatebetweenhardgelatin and soft gelatin capsules2.Compareandcontrast	Lectures Boards Slide show	Exams Seminars
10	3	Soft gelatin capsules	advantages and disadvantages of hard gelatin and soft gelatin capsules 3. List categories of inert ingredients, with examples, which are employed in themcompounding or manufacture of capsules 4. State USP compendial requirements for capsules	Lectures Boards Slide show	Exams Seminars
11	3	Pharmaceutical incompatabilities	1. Define pharmaceutical incompatabilities and list their types.	Lectures Boards Slide show	Exams Seminars

			2. Differentiate between their types		
12	3		<ol> <li>Define aerosols and list their types with brief description</li> <li>List the advantages of aerosols.</li> <li>List the main parts of aerosol system with their functions</li> <li>Describe each type of filling operation for aerosol.</li> <li>Differentiate between methods of filling</li> </ol>	Lectures Boards Slide show	Exams Seminars
13	3	Tablets	<ol> <li>Differentiate between the various types of tablet dosage forms.</li> <li>Compare and contrast advantages and disadvantages of the various types of tablet dosage forms</li> <li>List categories of ingredients, with examples, which are employed in the manufacture of compressed tablets.</li> </ol>	Lectures Boards Slide show	Exams Seminars
14	3		List the main procedures for tablet manufacturing	Lectures Boards Slide show	Exams Seminars
15	3		Discuss the main stability problems encountered in tablets dosage forms and describe their solutions	Lectures Boards Slide show	Exams Seminars

11. Course Evaluation			
Mid term exam 15+ quizzes and attendance 5-	+ practical 20+ 6	50 final exam	
12. Learning and Teaching Resources			
Required textbooks (curricular books, if any)			
Main references (sources)			
Recommended books and references			
(scientific journals, reports)			
Electronic References, Websites			

1. Course Name:

Physical pharmacy

2. Course Code:

### PhPpi2B00018

3. Semester / Year:

First semester/ second year

4. Description Preparation Date:

22/2/2024

5. Available Attendance Forms:

Class attendance

6. Number of Credit Hours (Total) / Number of Units (Total)

45credit hours /4 units

7. Course administrator's name (mention all, if more than one name)

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- 8. Course Objectives

8. Course Objectives				
Course Objectives	1-Identifying all types and forms of medicines.			
	2 - Methods of preparing active ingredients in full			
	dosage formsFor humans and animals			
	3 Study the stability of prepared doses in various			
	forms			
	4 - Study the drug effect, effectiveness, and			
	mechanism of action within the body.			
9. Teaching and Learning Strategies				
<b>Strategy</b> 1- Theoretical lectures				
2 - Educational laboratories				
3- Scientific and practical research				
1				
4- Desk research				
10. Course Structure				

Week	Hours	Required Learning	Unit or subject	Learning	Evaluation
		Outcomes	name	method	method
1-3	9		States of matter, binding forces between molecules, gases, liquids, solid and crystalline matters; phase equilibria and phase rule; thermal analysis		
1-2	6	Solutions of non- electrolytes	Solutions of non- electrolytes, properties, ideal and real colligative properties, molecular weight determination		

11. Course Evaluation							
The pursuit grade is divided into 20% for theory, 20% for practical, and 60% for the end- of-semester exam							
12. Learning and Teaching Resources							
Required textbooks (curricular books, if any)							
Main references (sources)							
Recommended books and references							
(scientific journals, reports…)							
Electronic References, Websites							