

First-Order Differential Equations

The general first-order differential equation is written as:

$$\frac{dy}{dx} = F(x, y) \quad \text{or} \quad M(x, y)dx + N(x, y)dy = 0$$

Separable Differential Equations: A first-order ODE is separable if it can be written in the form $f(x)dx = g(y)dy$ where the function $f(x)$ is independent of y and $g(y)$ is independent of x . We can find the general solution of this differential equation by integral

$$\int f(x)dx = \int g(y)dy$$

Example 1: Solve the ODE $xdy - 3ydx = 0$.

Solution: By separating the variables it becomes $\frac{dy}{y} = \frac{3dx}{x}$

$$\int \frac{dy}{y} = 3 \int \frac{dx}{x} \Rightarrow \ln y = 3 \ln x + \ln C \Rightarrow \ln y = \ln x^3 + \ln C$$

$$\ln y = \ln Cx^3 \Rightarrow y = Cx^3$$

Example 2: Find the general solution of the ODE $y' + 2e^{2x}y = e^{2x}y^2$.

Solution: From the given equation, we have $y' = (y^2 - 2y)e^{2x}$

$$\frac{dy}{dx} = (y^2 - 2y)e^{2x} \Rightarrow \frac{1}{y(y-2)} dy = e^{2x} dx$$

$$\frac{1}{y(y-2)} = \frac{A}{y} + \frac{B}{y-2}; \quad y=0 \Rightarrow A = -\frac{1}{2}, \quad y=2 \Rightarrow B = \frac{1}{2}$$

$$\int e^{2x} dx = \int \left[\frac{1}{2(y-2)} - \frac{1}{2y} \right] dy$$

$$\frac{1}{2} e^{2x} + C_1 = \frac{1}{2} \ln(y-2) - \frac{1}{2} \ln y$$

$$e^{2x} + 2C_1 = \ln(y-2) - \ln y \Rightarrow e^{2x} + C = \ln\left(\frac{y-2}{y}\right)$$

Example 3: Solve $\frac{dQ}{dx} + \frac{1}{2}Q = \frac{5}{2}$ with $Q(1) = 6$

Solution: We change to differential form, separate the variables, and integrate

$$\frac{dQ}{dx} = \frac{5}{2} - \frac{1}{2}Q \quad \Leftrightarrow \quad \frac{dQ}{dx} = -\frac{Q-5}{2}$$

$$\int_6^Q \frac{dy}{y-5} = -\frac{1}{2} \int_1^x dt$$

$$\ln(y-5) \Big|_6^Q = -\frac{1}{2}t \Big|_1^x$$

$$\ln(Q-5) - \ln 1 = -\frac{1}{2}x + \frac{1}{2} \quad \Leftrightarrow \quad \ln(Q-5) = \frac{1-x}{2}$$

$$Q-5 = e^{(1-x)/2} \quad \Leftrightarrow \quad Q = 5 + e^{(1-x)/2}$$

Example 4: Solve $e^x \tan y \, dx = (1 + e^x) \sec^2 y \, dy$ with $y(\ln 2) = \pi/4$

Solution:

$$\int_{\pi/4}^y \frac{\sec^2 u}{\tan u} du = \int_{\ln 2}^x \frac{e^t}{1 + e^t} dt \quad \Leftrightarrow \quad \ln(\tan u) \Big|_{\pi/4}^y = \ln(1 + e^t) \Big|_{\ln 2}^x$$

$$\ln(\tan y) - \ln(1) = \ln(1 + e^x) - \ln(1 + e^{\ln 2})$$

$$\ln(\tan y) = \ln(1 + e^x) - \ln(3)$$

$$\ln(\tan y) = \ln \frac{1 + e^x}{3}$$

$$\tan y = \frac{1 + e^x}{3}$$

$$y = \tan^{-1} \left(\frac{1 + e^x}{3} \right)$$

Newton's Low of Cooling: The rate of loss of heat from a body is directly proportional to the difference in the temperature of the body and its ambient.

Newton's law of cooling is given by: $\frac{dT}{dt} = -k(T - M)$

where t is time, k is the constant of proportionality, and M is the ambient temperature.

Example 5: Placed a metal bar, at a temperature of 40°C in a room with constant temperature of 0°C . After 20 minutes the temperature of the bar is 30°C . Determine the time required to reach the bar at a temperature of 20°C .

Solution: $M = 0^{\circ}\text{C} \Rightarrow \frac{dT}{dt} = -kT \Rightarrow \frac{dT}{T} = -kdt \Rightarrow \ln T = -kt + c_1$
 $T = e^{-kt+c_1} = e^{-kt} \cdot e^{c_1}$

$$\boxed{T = ce^{-kt}} ; (e^{c_1} = c)$$

(Initial condition) $t = 0, \quad T = 40^{\circ}\text{C}$

$$40 = ce^{-k \times 0} \Rightarrow c = 40$$

$$T = 40e^{-kt}$$

To find k we have $t = 20 \text{ min}$, $T = 30^{\circ}\text{C} \Rightarrow 30 = 40e^{-k \times 20}$

$$e^{-20k} = 0.75 \Rightarrow -20k = \ln(0.75)$$

$$\therefore k = \frac{\ln(0.75)}{-20} = 0.0144$$

$$T = 40e^{-0.0144t}$$

$$20 = 40e^{-0.0144t} \Rightarrow e^{-0.0144t} = 0.5 \Rightarrow t = \frac{\ln(0.5)}{-0.0144} = 48 \text{ minutes.}$$

Example 6: Water at a temperature of 80°C is placed in a room which is held at a constant temperature of 25°C . How much would be the temperature of water after 10 minutes if $k = 0.056$.

Solution: $\frac{dT}{dt} = -0.056(T - 25) \Rightarrow \frac{dT}{(T - 25)} = -0.056dt$
 $\ln(T - 25) = -0.056t + c_1 \Rightarrow T - 25 = e^{-0.056t+c_1}$

$$T = 25 + ce^{-0.056t} ; c = e^{c_1}$$

$$t = 0 , \quad T = 80 \Rightarrow 80 = 25 + c \Rightarrow c = 55$$

$$T = 25 + 55e^{-0.056t} \Rightarrow T(10) = 25 + 55e^{-0.056 \times 10} = 56.4^{\circ}\text{C}$$

Radioactive Decay: Radioactive decay is the random process in which a nucleus loses energy by emitting radiation. The number of decays per second is given by:

$$\frac{dN}{dt} = \lambda N$$

where λ is the decay constant and N represent to the number of undecayed nuclei. The half-life $T_{1/2}$ of a sample is the time taken for half of the radioactive nuclei to decay:

$$T_{1/2} = -\frac{\ln 2}{\lambda}$$

Example 7: The half-life of radioactive radium is 1600 years. If a sample initially contains 50 gm, how long will it be until it contains 45 gm?

Solution:

$$T_{1/2} = -\frac{\ln 2}{\lambda} \quad \Leftrightarrow \quad 1600 = -\frac{\ln 2}{\lambda} \quad \Leftrightarrow \quad \lambda = -\frac{\ln 2}{1600}$$

$$\frac{dN}{dt} = \lambda N \quad \Leftrightarrow \quad \frac{dN}{N} = \lambda dt$$

$$\int \frac{dN}{N} = \int \lambda dt \quad \Leftrightarrow \quad \ln N = \lambda t + C$$

$$N = Ae^{\lambda t}; A = e^C$$

We have $N = 50$ when $t = 0 \quad \Leftrightarrow \quad 50 = Ae^0 \quad \Leftrightarrow \quad A = 50$

Thus $N = 50e^{\lambda t}$

$$\text{So, } N = 50e^{-\frac{\ln 2}{1600}t} \quad \Leftrightarrow \quad \frac{N}{50} = e^{-\frac{\ln 2}{1600}t}$$

$$\frac{45}{50} = e^{-\frac{\ln 2}{1600}t} \quad \Leftrightarrow \quad -\frac{\ln 2}{1600}t = \ln\left(\frac{45}{50}\right)$$

$$t = -\frac{1600 \ln(0.9)}{\ln 2} = 243.2 \quad \text{years}$$

H.W

Solve the ODEs:

(1) $x(2y - 3)dx + (x^2 + 1)dy = 0$ (Ans. $(x^2 + 1)(2y - 3) = K$)

(2) $t^2 dT + tT dt = (t + 6)dT + 2T dt$. (Ans. $T^5(t + 2)^4(t - 3) = K$)

(3) $(1 + u^2)dv - (1 + v^2)du = 0$ with $v(0) = 1$

(4) $\frac{ds}{dt} = \frac{s^2 - s - 2}{t^2 + t}$ with $s(1) = 3$

(5) A body at a temperature of 40°C is placed in a room with constant temperature of 20°C . If after 10 minutes the temperature of the body is 35°C , find the time required for the body to reach a temperature of 30°C . (Ans: 14.096 minutes)

(6) The half-life of radioactive einsteinium is 276 days. After 100 days, 0.5 gram remains. What was the initial amount?

(7) The half-life of radioactive radium is 1600 years. If a sample initially contains 30 grams, how much the amount will remain after 250 years? (Ans: 26.9 grams)

Websites:

1. <https://tutorial.math.lamar.edu/Classes/DE/Separable.aspx>
2. https://www.sfu.ca/math-coursenotes/Math%20158%20Course%20Notes/sec_first_order_homogeneous_linear.html