

Republic of Iraq
The Ministry of Higher
Education and Scientific
Research University
of Babylon
College of Science for Women



جمهورية العراق
وزارة التعليم العالي والبحث العلمي
جامعة بابل
كلية العلوم للبنات

alkaimayad@gmail.com

العدد: 2

Nanotechnology

رقم المقرر الدولي:

التاريخ: April 2020

استمارة خاصة لاطلاع الطلبة على المقرر

Course NO.:

1	اسم المقرر	Nanotechnology	النوع	فصلي	إجباري
2	عدد الوحدات	3	عدد الساعات الإجمالي	3	النظري
				3	العملي
3	المرحلة الدراسية	ماستر			
4	اسم التدريسي	اياد فاضل محمد القيم alkaimayad@gmail.com			

الشهادة واللقب العلمي	دكتوراه - أستاذ
التخصص	الكيمياء الفيزيائية

5	لغة تدريس المقرر	انكليزي
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برنامج المقرر

اولا	المقدمة
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Lectures: All lectures will support the post graduate students to learn the general basic of materials and synthesis of them as a nanomaterial starting from simple method for preparation and going to top highly techniques for this topic.

Theoretical exercises and applications:. Post graduate student shall be capable to learn the XRD, TEM, SEM, UBVis band-gap and materials structures.

Learning Objectives

After careful study of this course the student should be able to:

- I- Type of materials .
- II- Macro and nano-materials behaviors.
- III- Adsorption of surface
- IV- role of nanomaterials in our life .
- V- How to calculate the crystal size and particle size .
- VI- Role of these materials in industry.
- VII- Role of the visible light and UV lights

المقدمة		أولا
مفردات المقرر او المحتوى		ثالثا
Week 1	Introduction Nanotechnology	
Week 2	Role of the surface by depend on the semiconductor	
Week 3	Role of metals doping	
Week 4	Adsorption of the surface and models	
Week 5	Photolysis and photofenton	
Week 6	Photo catalysis	
Week 7	Exam no. 1	
Week 8	Role of lights and air	
Week 9	Role of techniques :	
Week 10	XRD, UV-Band Gap	
Week 11	XPS, EDX	
Week 12	AFM, TEM and SEM	
Week 13	Role of Langmuir Hanshilwood model	
Week 14	Exam no. 2	
طرائق التدريس المعتمدة لتنفيذ المقرر		رابعا
1-Zoom program		
2-Free call		
3-Google classroom,		
4-work shops		

اولا	المقدمة
خامسا	توزع درجات المقرر وفق لوائح أو أكثر مما يأتي
16 %	<ul style="list-style-type: none"> • Active participation, homework assignments, Attendance, quizzes
34%	<ul style="list-style-type: none"> • Do more than 2 formal exams either online or handbook
50%	<ul style="list-style-type: none"> • Preparing material of seminars as a reports
سادسا	المصادر والمراجع
<p>1) Nanomaterials: Biomedical, Environmental, and Engineering Applications By: Part I: Nanomaterials: Synthesis and Characterization 1 “Synthesis, Characterization and General Properties of Carbon Nanotubes 3 Falah H. Hussein, Firas H. Abdulrazzak, and Ayad F. Alkaim/ 2018”</p> <p>https://onlinelibrary.wiley.com/doi/10.1002/9781119370383.ch1</p> <p>2) Nanostructures and Nanomaterials: Synthesis, Properties, and Applications (2nd Edition) (World Scientific Series in Nanoscience and Nanotechnology) 2nd ed. Edition/2019</p> <p>3) Lecture notes of Nanomaterial/ 2020</p>	
<p>مدرس المادة أ.د. اياد فاضل محمد القيم</p> <p>التوقيع:</p> <p>رئيس القسم:</p>	

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Mohammed29121971@gmail.com

العدد:	Application of Bonding Theories in Coordination compounds	رقم المقرر الدولي:
التاريخ: April 2023	استمارة خاصة لاطلاع الطلبة على المقرر	Course NO.:

إجباري	فصلي	النوع	Application of Bonding Theories in Coordination compounds	اسم المقرر	1
	العملي	٣	النظري	٣	عدد الوحدات
			عدد الساعات الإجمالي	٣	2
			ماستر	المرحلة الدراسية	3
			محمد حامد سعيد الدهيمي	اسم التدريسي	4
			Mohammed29121971@gmail.com		

الشهادة واللقب العلمي	دكتوراه - أستاذ
التخصص	الكيمياء اللاعضوية

5	لغة تدريس المقرر	انكليزي
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برنامج المقرر

اولا	المقدمة
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Coordination Chemistry provides the information on the mode of action of biological molecules based on their structural studies. With such knowledge, scientists have been able to design and modify several important biological molecules. This course will expose students to the applications of previous knowledge acquired in spectroscopic techniques, Chemical kinetics, thermodynamics and reaction mechanisms. The course will open a new world to students with interest in medicine, where knowledge of coordination chemistry is employed in the extraction of metal poison in living systems, and petrochemical industry, where the knowledge of coordination chemistry is important in catalyst design and application

Learning Objectives

At the end of this course, the students should be able to:

- ☐ Identify coordination compounds, explain the methods used in preparing them and state areas of their applications.
- ☐ Name, classify and identify the possible number of isomers of any given coordination compounds.
- ☐ Describe the structures and hybridizations of coordination compounds.
- ☐ Apply physical techniques in characterization of coordination compounds.
- ☐ Explain the nature of bonding in coordination compounds through the various bonding theories.
- ☐ Apply the knowledge of coordination chemistry in stabilization of unusual oxidation states.
- ☐ State the theories used to describe bonding in metal complexes
- Explain Valence Bond Theory (VBT)

المقدمة		اولا
<ul style="list-style-type: none"> • Explain Crystal Field theory (CFT) • Explain Ligand Field Theory (LFT) • Explain Molecular Orbital Theory (MOT) 		
مفردات المقرر او المحتوي		ثالثا
Week 1	Introduction Coordination Compounds	
Week 2	Nomenclature of Coordination Compounds	
Week 3	Ligand , Type of ligand	
Week 4	Chain Theory and Werner's Theory	
Week 5	Sidgwick Theory Effective Atomic Number Rule	
Week 6	Hybridization of Atomic Orbitals	
Week 7	Valence Bond Theory , postulates of v.B. theory	
Week 8	Hybrid orbitals and bonding in the octahedral	
Week 9	Hybrid orbitals and bonding in the square planar, Hybrid orbitals and bonding in the tetrahedral	
Week 10	Crystal field theory, Crystal field splitting in octahedral complexes	
Week 11	Factors Affecting Crystal Filed Splitting, Limitations of CFT	
Week 12	Ligand Field or Molecular Orbital Theory	
Week 13	Molecular orbital theory	
Week 14	Electronic spectra	
Week 15	Exam	

اولا	المقدمة
رابعا	طرائق التدريس المعتمدة لتنفيذ المقرر
1-Use the Smart screen as a modern Academy classroom facilities 2-Power Point Data Show presentations, 3-Concentrating on the activities of the students 4-Assessments	
خامسا	توزيع درجات المقرر وفق لوائح أو أكثر مما يأتي
16 % 34% 50%	<ul style="list-style-type: none"> • Active participation, homework assignments, Attendance, quizzes • Do more than 2 formal online exams • Solving a several related questions
سادسا	المصادر والمراجع
<u>The References:-</u> 1- Advanced Inorganic Chemistry by F.Albert Cotton ,Geoffrey Wilkinson , Carlos A. Murillo , Manfred Bochmann ; Sixth Edition 2009 2- Concise Inorganic Chemistry by J.D.Lee ; Fifth Edition 2011 3-Introduction to Coordination Chemistry by Geoffrey A. Lawrance ; First Edition 2009 4- Inorganic Chemistry principles of structure and reactivity by James E. Huheey , Ellen A. Keiter , Richard L. Keiter , Okhil K. Medhi ; Fourth Edition 2006	
مدرس المادة أ.د. محمد حامد سعيد الدهيمي	التوقيع:

اولا	المقدمة
رئيس القسم ومدرس المادة:	

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العدد: 2	Advances in Electrochemistry	رقم المقرر الدولي:
التاريخ: April 2020	استمارة خاصة لاطلاع الطلبة على المقرر	Course NO.:

1	اسم المقرر	Advances in Electrochemistry	النوع	فصلي	إجباري
2	عدد الوحدات	3	عدد الساعات الإجمالي	3	النظري
				3	العملي
3	المرحلة الدراسية	ماستر			

حازم يحيى محمد علي الجبوري h.yahya40@yahoo.com	اسم التدريسي	4
دكتوراه - أستاذ مساعد	الشهادة واللقب العلمي	
الكيمياء الفيزيائية	التخصص	

انكليزي	لغة تدريس المقرر	5
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برنامج المقرر

المقدمة	اولا
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Lectures: The lectures will present the general chemical background of the measurement principles and instrumental techniques as well as theory on sampling, data processing, data analysis, quality assurance and method validation.

Theoretical exercises: These exercises will train the students to carry out calculations on the data produced from different instrumental techniques including statistical analysis.

Learning Objectives

After careful study of this course the student should be able to:

- VIII- List and describe the conductors .
- IX- Explain the meaning of the molar conductors.
- X- Define conductors unit, specific conductors, molar and equivalent conductors.
- XI- Calculate the molar conductors .
- XII- Explain the strong and weak electrolyte and the effect of concentration on the molar conductors.
- XIII- Derive and use the Nernst equation.
- XIV- Define and calculate the free energy.

المقدمة		اولا
XV-	Describe the general procedure for calculation of transport number.	
XVI-	Describe the types of electrochemical cell.	
مفردات المقرر او المحتوي		ثالثا
Week 1	Introduction to the Electrochemistry	
Week 2	What are the molar and equivalent conductors?	
Week 3	Effect of dilution on molar conductance of strong and weak electrolyte	
Week 4	Transport number of ions?	
Week 5	Ostwald law of dilution	
Week 6	Electrochemical cell?	
Week 7	Galvanic cell and electrolytic cell?	
Week 8	Standard half-cell potentials.	
Week 9	Cell potentials and free energy.	
Week 10	<ul style="list-style-type: none"> The Nernst equation. Concentration cells.	
Week 11	<ul style="list-style-type: none"> Analytical applications of the Nernst equation. Determination of solubility products	
Week 12	<ul style="list-style-type: none"> Potentiometric titrations. Measurement of pH.	
Week 13	Batteries and fuel cells.	
Week 14	<ul style="list-style-type: none"> Electrochemical Corrosion. Control of corrosion.	

المقدمة		اولا
Week 15	Exam	
طرائق التدريس المعتمدة لتنفيذ المقرر		رابعا
1-Use the Smart screen as a modern Academy classroom facilities 2-Power Point Data Show presentations, 3-Concentrating on the activities of the students 4-Assessments		
توزيع درجات المقرر وفق لوائح أو أكثر مما يأتي		خامسا
<ul style="list-style-type: none"> Active participation, homework assignments, Attendance, quizzes Do more than 2 formal online exams Solving a several related questions 		16 %
		34%
		50%
المصادر والمراجع		سادسا
[1] Physical Chemistry, Atkins, 6 th ed. 2001. [2] Problems in physical chemistry 1 st , by K.K. Shurma, 1994. [3] Physical chemistry 5 th by Walter J. Moor, 1972. Physical chemistry 7 th by Robert Al-Berty, 1987.		
التوقيع:		مدرس المادة أ.م.د. حازم يحيى الجبوري
رئيس القسم ومدرس المادة:		

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Sadiqkarim77@gmail.com

العدد: 10	Advances of Organic Spectroscopy	رقم المقرر الدولي:
التاريخ: April 2020	استمارة خاصة لاطلاع الطلبة على المقرر	Course NO.:

إجباري	فصلي	النوع	Advances of Organic Spectroscopy	اسم المقرر	1
	العملي	٣	النظري	٣	عدد الوحدات
			عدد الساعات الإجمالي	٣	2
			ماستر	المرحلة الدراسية	3
			صادق عبدالحسين كريم	اسم التدريسي	4
			Sadiqkarim77@gmail.com		

الشهادة واللقب العلمي	دكتوراه - أستاذ مساعد
التخصص	الكيمياء – الكيمياء العضوية والبوليمرات

5	لغة تدريس المقرر	انكليزي
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برنامج المقرر

اولا	المقدمة
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Lectures: The subject will present the general chemical background of the The study of molecular structure and dynamics through the absorption, emission and scattering of light.

Theoretical exercises: These exercises will train the students to carry out investigation of organic compounds which the organic molecules are in a constant state of vibrations, each bond having its characteristic stretching and bending frequencies

Learning Objectives

After careful study of this course the student should be able to:

- XVII- List of concept of UV-Vis.
- XVIII- The Study of effect the UV-Vis of organic compounds.
- XIX- List of concept of FTIR.
- XX- The Study of effect the FTIR of organic compounds.
- XXI- List of concept of HNMR.
- XXII- The Study of effect the HNMR of organic compounds.
- XXIII- List of concept of ¹³CNMR.
- XXIV- The Study of effect the ¹³CNMR of organic compounds.
- XXV- List of concept of Mass spectroscopy.
- XXVI- The Study of effect the Mass spectroscopy of organic compound

ثالثا	مفردات المقرر او المحتوي
	Week 1 List of concept of UV-Vis.

المقدمة		اولا
Week 2	The Study of effect the UV-Vis of organic compounds	
Week 3	List of concept of FTIR	
Week 4	The Study of effect the FTIR of organic compounds	
Week 5	The Study of effect the FTIR of organic compounds	
Week 6	The Study of effect the FTIR of organic compounds	
Week 7	List of concept of HNMR	
Week 8	The Study of effect the HNMR of organic compounds	
Week 9	The Study of effect the HNMR of organic compounds	
Week 10	The Study of effect the HNMR of organic compounds	
Week 11	List of concept of ¹³ CNMR.	
Week 12	The Study of effect the ¹³ CNMR of organic compounds	
Week 13	List of concept of Mass spectroscopy	
Week 14	The Study of effect the Mass spectroscopy of organic compounds.	
Week 15	Exam	
طرائق التدريس المعتمدة لتنفيذ المقرر		رابعا
1-Use the Smart screen as a modern Academy classroom facilities 2-Power Point Data Show presentations, 3-Concentrating on the activities of the students 4-Assessments		

اولا		المقدمة
خامسا		توزع درجات المقرر وفق لواحد أو أكثر مما يأتي
16 %		• Active participation, homework assignments, Attendance, quizzes
34%		• Do more than 2 formal online exams
50%		• Solving a several related questions
سادسا		المصادر والمراجع
1- P.Crews, J.rodriquez , M.jaspars; Organic structure analysis (1998).		
2- J.R.dyer; Applications of absorption spectroscopy of organic compounds;(1965).		
3- F.Scheinmann ; Introduction to spectroscopic methods; (1973).		
4- L.Lang ; Absorption spectra in the UV&VIS. Region ; (1995)		
<p>التوقيع: _____</p> <p>مدرس المادة أ.م.د. صادق عبدالحسين كريم</p> <p>رئيس القسم ومدرس المادة: _____</p>		

جمهورية العراق

وزارة التعليم العالي والبحث العلمي

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fouadalkaim@yahoo.com

العدد: 2	Liquid chromatography application	رقم المقرر الدولي:
التاريخ: April 2020	استمارة خاصة لاطلاع الطلبة على المقرر	Course NO.:

1	اسم المقرر	Liquid chromatography application	النوع	فصلي	إجباري
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2	عدد الوحدات	2	عدد الساعات الإجمالي	2	النظري	2	العملي
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3	المرحلة الدراسية	ماجستير
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4	اسم التدريسي	فؤاد فاضل القيم fouadalkaim@yahoo.com
	الشهادة واللقب العلمي	دكتوراه - أستاذ مساعد
	التخصص	الكيمياء التحليلية

5	لغة تدريس المقرر	انكليزي
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برنامج المقرر

اولا	المقدمة
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It could be more interested for those are willing to analysis the organic compounds in water and other real samples.

This subject will cover many thing related to high performance liquid chromatography theoretically and practically.

Objectives

- XXVII- Introducing the HPLC instrument.
- XXVIII- Explaining the parts of HPLC.
- XXIX- Role of the detector used.
- XXX- Study the properties of columns in terms of separation.
- XXXI- Calculation the resolution, separation factor and so on
- XXXII- Define the *van Deemter* equation
- XXXIII- Recognition between retained and un-retained peaks.
- XXXIV- Dealing with real sample analysis.

مفردات المقرر او المحتوي		ثالثا
Week 1	Definition of chromatographic separation	
Week 2	Describe the parts of HPLC: solvent reservoir and its properties	

المقدمة		اولا
Week 3	Describe the parts of HPLC: Pumps and sample injector	
Week 4	Elution program system	
Week 5	Describe the parts of HPLC: columns	
Week 6	Describe the parts of HPLC: detectors	
Week 7	How to separate two components in HPLC?	
Week 8	Why chromatographic peak does become broad?	
Week 9	Monthly examination 1	
Week 10	Separation factor	
Week 11	Resolving power	
Week 12	Solution of problems	
Week 13	Monthly examination 2	
Week 14	Analysis of emergent contaminants using HPLC/LC-MS	
Week 15	Method validation	
طرائق التدريس المعتمدة لتنفيذ المقرر		رابعا
1-Google class room 2-Power Point Data Show presentations with sound 3-Daily homework 4-Whatsapp group for more communication		

اولا	المقدمة
خامسا	توزع درجات المقرر وفق لواحد أو أكثر مما يأتي
10 %	• Active participation, homework and Attendance
40%	• Quizzes and assignments
50%	• Monthly examination
سادسا	المصادر والمراجع
Fundamentals of analytical chemistry, fourth edition, Douglas A. Skoog, Stanford University	
<p>مدرس المادة أ.م.د. فؤاد فاضل القيم</p> <p>التوقيع:</p> <p>رئيس القسم ومدرس المادة:</p>	

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Talat.tariq@yahoo.com

العدد: 2

Bionanoparticles

رقم المقرر الدولي:

التاريخ: April 2020

استمارة خاصة لاطلاع الطلبة على المقرر

Course NO.:

1	اسم المقرر	Bionanoparticles	النوع	فصلي	إجباري
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2	عدد الوحدات	3	عدد الساعات الإجمالي	3	النظري	3	العملي
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3	المرحلة الدراسية	ماستر
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4	اسم التدريسي	طلعت طارق خليل الشمري Talat.tariq@yahoo.com
	الشهادة واللقب العلمي	دكتوراه - أستاذ مساعد
	التخصص	الكيمياء الحياتية

5	لغة تدريس المقرر	انكليزي
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برنامج المقرر

اولا	المقدمة
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Lectures: The lectures will explain the nanoparticlesl background and methodology of nanoparticle as well as bionanoparticles, then classification of nanoparticles and bionanoparticles using for diagnosis and treatment in diseases.

Theoretical exercises: These exercises will ask the students to know different between micronanoparticles

and nanoparticles . using of nanoparticles is in medicin

Learning Objectives

After careful study of this course the student should be able to:

- XXXV- List and describe the nanoparticles .
- XXXVI- Explain the meaning of the nanoparticles .
- XXXVII- Classification of nanoparticles
- XXXVIII- Explain different between Micronanoparticles and nanoparticles
- XXXIX- Define Bionanoparticle
- XL- Types of Bionanoparticles
- XLI- Explain using of nanoparticles for treatment and diagnosis in diseases
- XLII- Toxicology of nanoparticles

مفردات المقرر او المحتوي

ثالثاً

Week 1	Introduction to the Nanoparticles
Week 2	Classification of nanoparticles?
Week 3	What are difference between Micronanoparticles and nanoparticles?
Week 4	Using of nanoparticles.
Week 5	Methodology for preparation of nanoparticles .
Week 6	Physical methods.
Week 7	Chemical methods.
Week 8	Definition of Bionanoparticles
Week 9	History of Bionanoparticles

المقدمة		اولا
Week 10	Classification of Bionanoparticles	
Week 11	Using of Bionanoparticles for diagnosis of disease	
Week 12	Using of Bionanoparticles for treatment of disease	
Week 13	Methodology for estimation of bionanoparticles in medicine	
Week 14	Toxicology of nanoparticles	
Week 15	Exam	
طرائق التدريس المعتمدة لتنفيذ المقرر		رابعا
1-Use the Smart screen as a modern Academy classroom facilities 2-Power Point Data Show presentations, 3-Concentrating on the activities of the students 4-Assessments		
توزيع درجات المقرر وفق لوائح أو أكثر مما يأتي		خامسا
<ul style="list-style-type: none"> Active participation, homework assignments, Attendance, quizzes Do more than 2 formal online exams Solving a several related questions 		16 %
		34%
		50%
المصادر والمراجع		سادسا
[1] Handbook of Nanoparticles, Mahmood Aliofkhazraei, 2015.		

اولا	المقدمة
[2] Nanobiomaterials Handbook, <i>Balaji Sitharaman 1st d</i> , 2017.	
[3] The delivery of nanoparticles, Abbass A. Hashim, 2012.	
<p>التوقيع: _____</p> <p>مدرس المادة أ.م.د. طلعت طارق خليل</p> <p>رئيس القسم ومدرس المادة: _____</p>	

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annenayad@gmail.com

العدد: 2	Extraction	رقم المقرر الدولي:
التاريخ: April 2020	استمارة خاصة لاطلاع الطلبة على المقرر	Course NO.:

1	اسم المقرر	Extraction				النوع	فصلي	إجباري
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2	عدد الوحدات	3	عدد الساعات الإجمالي	3	النظري	3	العملي	
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3	المرحلة الدراسية	ماستر						
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4	اسم التدريسي	اسيل مشتاق كاظم annenayad@gmail.com						
	الشهادة واللقب العلمي	دكتوراه - أستاذ مساعد						
	التخصص	الكيمياء التحليلية						

5	لغة تدريس المقرر	انكليزي						
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برنامج المقرر								
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اولا	المقدمة							
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<p>Lectures: All lectures will support the post graduate students to learn the general basic of analytical chemistry and its role for sharing with Physical chemistry applications, this topic will rehabilitate the post graduate student to have a new terms of extractions in solid/liquid and a liquid-liquid.</p> <p><u>Learning Objectives</u></p> <p><u>After careful study of this course the student should be able to:</u></p> <p>XLIII- Extraction by a solvent</p>								
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المقدمة		اولا
XLIV- The different between distribution constant and percentage XLV- How to do a fast separation of materials XLVI- Type of mixtures XLVII- Type of adsorption XLVIII- How to control on the pollutants XLIX- Types of dyes		
مفردات المقرر او المحتوى		ثالثا
Week 1	Introduction of Extraction	
Week 2	Liquid-Liquid Extraction	
Week 3	Distribution constant and effect of temperature	
Week 4	Partition coefficient	
Week 5	Selectivity of the extraction	
Week 6	Role of thermodynamic parameters in Extraction.	
Week 7	Exam no. 1	
Week 8	Nature of partition force	
Week 9	Separation factors	
Week 10	Adsorption	
Week 11	Models of adsorption	
Week 12	Solvent extraction systems	
Week 13	Dyes and it is removal	
Week 14	Exam no. 2	

المقدمة		اولا
طرائق التدريس المعتمدة لتنفيذ المقرر		رابعا
1-Zoom program 2-Free call 3-Google classroom, 4-work shops		
توزع درجات المقرر وفق لواحد أو أكثر مما يأتي		خامسا
<ul style="list-style-type: none"> Active participation, homework assignments, Attendance, quizzes Do more than 2 formal exams either online of handbook Preparing material of seminars as a reports 		16 %
		34%
		50%
المصادر والمراجع		سادسا
4) Analytical Chemistry / Skoog/ Ed. 7 2010 5) Ionic Liquid as Novel Solvent for Extraction and Separation in Analytical Chemistry By Li Zaijun, Sun Xiulan and Liu Junkang/ 2011 6) Lecture notes of Extractions/ 2020 7) Adamson/ surface chemistry/ ed.4/ 1978		
التوقيع:		مدرس المادة أ.م. د. أسيل مشتاق الجبوري

المقدمة	أولا
رئيس القسم:	